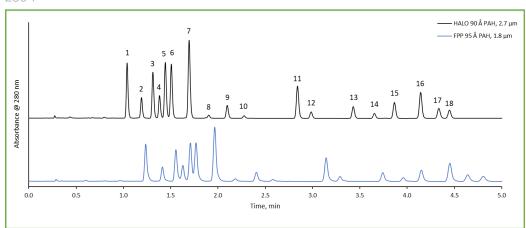


ENVIRONMENTAL



Comparison of HALO® PAH vs. FPP column for 18 PAH Compounds





PEAK IDENTITIES

- 1. Naphthalene
- 2. Acenaphthylene
- 3. 1-methylnaphthalene
- 4. 2-methylnaphthalene
- 5. Acenaphthene
- 6. Fluorene
- 7. Phenanthrene
- 8. Anthracene
- 9. Fluoranthene
- 10. Pyrene
- 11. Benzo(a)anthracene
- 12. Chrysene
- 13. Benzo[b]fluoranthene
- 14. Benzo[k]fluoranthene
- 15. Benzo[a]pyrene
- 16. Dibenzo[a,h]anthracene
- 17. Benzo[g,h,i]perylene
- 18. Indeno[1,2,3-cd]pyrene

TEST CONDITIONS:

Column: HALO 90 Å PAH, 2.7 μm, 4.6 x 50 mm

Competitor Column: FPP 95 Å PAH, 1.8 µm, 4.6 x 50 mm

Part Number: 92844-412 Mobile Phase A: Water

B: Acetonitrile

50

Gradient: Time %B 0.0 50 4.0 100 5.0 100

Flow Rate: 1.8 mL/min

HALO® Back Pressure: 256 bar Competitor Back Pressure: 344 bar

5.01

Temperature: 30 °C
Detection: 280 nm
Injection Volume: 2 µL
Sample Solvent: Methanol
Data Rate: 100 Hz

Response Time: 0.025 sec

Flow Cell: 1 µL

LC System: Shimadzu Nexera

Polycyclic Aromatic Hydrocarbons (PAHs) are a group of more than 100 chemicals released from the combustion of coal, oil, gasoline, tobacco, and wood. They can also be found in cooked food. PAHs are persistent chemicals and must be closely regulated for early detection/monitoring to minimize hazardous exposure in the environment and/or use of contaminated raw materials in different industries. A separation of eighteen PAH compounds is performed on a HALO® PAH column and a FPP PAH competitor column. The HALO® column shows excellent peak resolution, along with a lower overall back pressure compared to the competitor's unresolved peaks and peak tailing.



