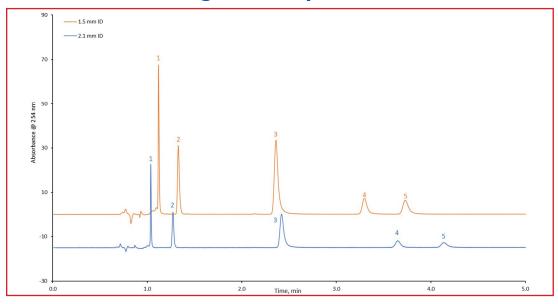
CLINICAL / TOXICOLOGY



Nicotine Metabolite Comparison of 1.5 mm to 2.1 mm ID Column using HILIC Separation Mode



TEST CONDITIONS:

314

Column: HALO 90 Å Penta HILIC, 2.7 µm 1.5 x 150 mm

Part Number: 9281X-705

Column: HALO 90 Å Penta HILIC, 2.7 μ m 2.1 x 150 mm

Part Number: 92812-705

Isocratic: 10/90 20mM Ammonium Formate @ pH=3/

ACN + 0.1% Formic Acid

Flow Rate: 0.4 mL/min Pressure: 188 Bar - 1.5 mm

73 Bar - 2.1 mm

Temperature: 40 °C

Detection: UV 254 nm, PDA **Injection Volume:** 1.0 µL

Sample Solvent: 10/90 Water/ACN

Data Rate: 100 Hz

Response Time: 0.025 sec.

Flow Cell: 1 µL

LC System: Shimadzu Nexera X2

PEAK IDENTITIES

- 1. Cotinine
- 2. Trans-3-hydroxycotinine
- 3. Nicotine
- 4. Anabasine
- 5. Nornicotine

Nicotine is a widely known alkaloid found in cigarettes and is highly addictive. In order to break the addiction to nicotine, people must be weaned off the alkaloid slowly to reduce the side effects of withdrawal. This requires monitoring of nicotine and its metabolites as well. Due to the basic nature of nicotine, it is difficult to separate under reversed phase conditions. By running the samples under HILIC conditions, there is sufficient retention for each chemical to be separated completely. In the above chromatogram a comparison of the 1.5 and 2.1mm ID's is shown. The HALO® 1.5 mm ID Penta-HILIC column produces a similar separation but with increased peak heights and increased area counts for the same injection volume of sample. When working with clinical samples that can be low in abundance, a 1.5 mm ID column can produce a lower LOD for problematic samples.



