Method Development Approach to Separating Oligonucleotides Under UV and MS Detection

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Introduction to Oligos

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- Why are oligonucleotides being analyzed?
 - Oligonucleotides are being developed as drugs
 - These drugs can be used to modulate gene expression
 - Impurity analysis of said drugs is required for effective products
 - The development of mRNA vaccines (Covid-19)
- What are the challenges of separating oligonucleotides?
 - Complicated structures and sizes
 - A phosphate backbone that reduces retention under RPLC
 - Modifications can range broadly: phosphorothioates, lipids, sugars or nucleobases
 - Sample stability requires suitable handling (nucleases and metals)
 - Hybrid structure formation can be either intended or unwanted

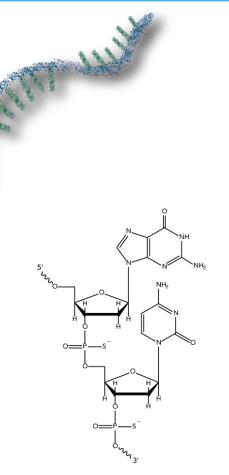


Figure 1: Structure of a 2'-O-modified RNA. (idtdna.com)



Technical Considerations for Oligonucleotide Separations

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- Reversed-phase or ion-pairing RP HPLC methods have largely replaced gel electrophoretic methods for small (<20 nts) and medium size (<60 nts) oligo/poly nucleotides.
- IP-RP methods can be rapid, and with suitable choices can be used with online MS detection for identity and sequence analysis.
- Most IP-RP approaches are at pH 6-11, and often at elevated temperature, especially when complementary hybrids can be formed (internal or intermolecular hybrids).
- Hybrid formation is temperature, ionic strength and solvent dependent, and defined by sequence (GC vs AT or AU), as well as backbone (RNA vs DNA).

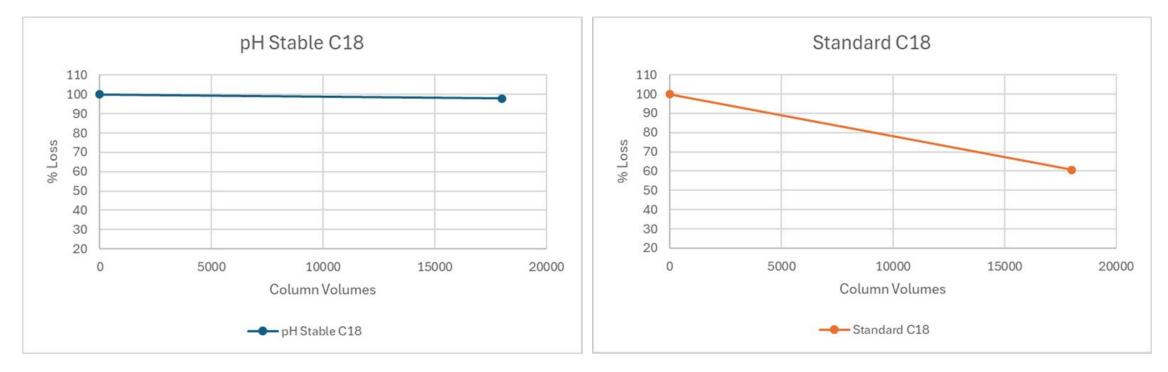


Importance of High pH Stability

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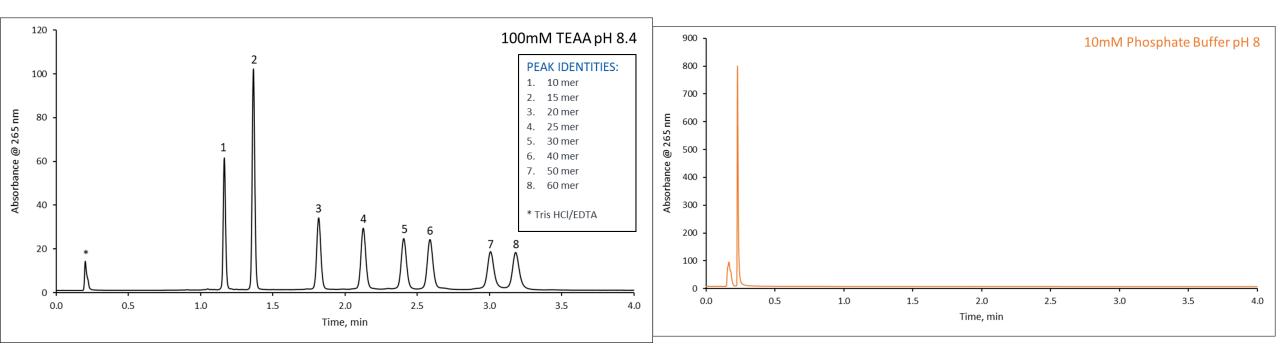
- At high pHs, native silica will be solubilized unless treated for high pH stability
- The graph measures the result of efficiency (plates) of both a standard silica and the new surface modified silica from AMT after stability testing.
- The pH stable phase maintained good efficiency after the stability while a standard, silica particle bonded phase material, lost efficiency.



Why Ion Pairing?

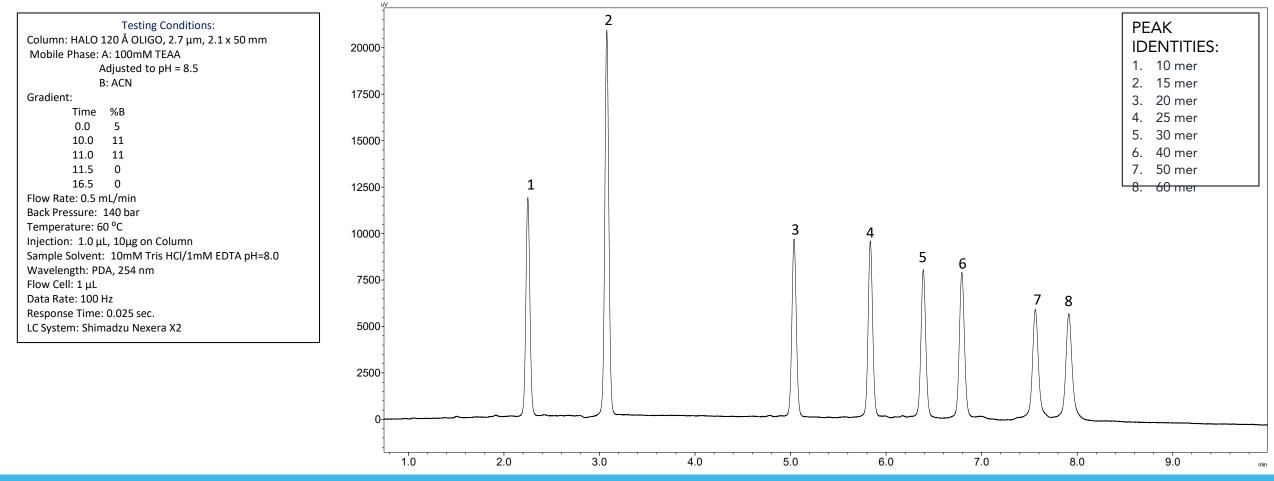
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- Contemporary oligonucleotide analysis is frequently conducted using ion pair reversed phase chromatography (IPRP).
- TEA (triethylamine) or TEA-Acetate (TEAA) buffer is a common IPRP modifier, creating favorable IPRP separations.
- Phosphate buffer does not support IPRP, meaning little to no retention.





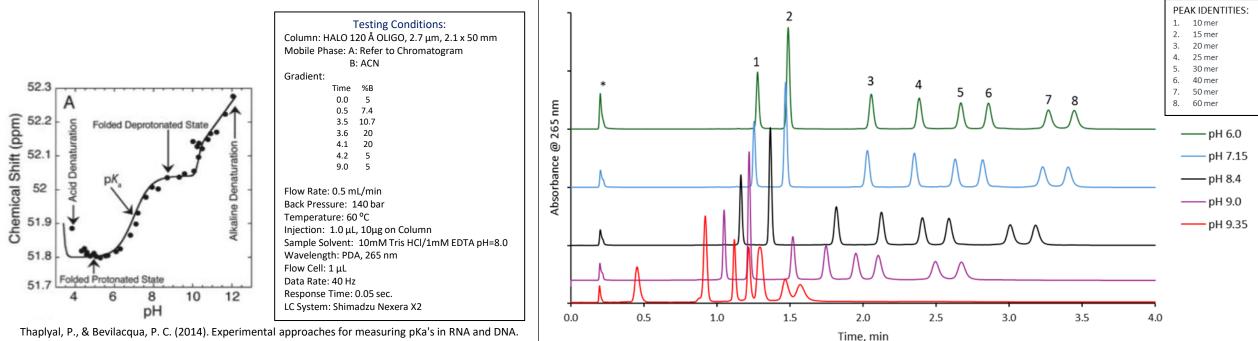
Standard Separation with Ion Pairing





Effects of pH on Oligonucleotides

- Five different pHs were used to evaluate how the retention of oligonucleotides differs at the pHs 6 and 9.5
- TEAA was used as the ion pairing reagent for each mobile phase and acetic acid was used ٠ to adjust the pH as specified
- The concentration of TEA was maintained at 100mM to ascertain the role that pH has on ٠ oligonucleotide retention
- By adjusting pH with acetic acid, we can see, retention modestly decreases as pH increases ٠
- Around a pH of 9, it is suggested that at 60°C oligonucleotides will adopt a folded ٠ deprotonated state, reducing retention

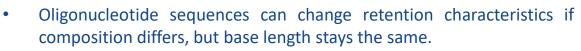


Methods in enzymology, 549, 189-219. https://doi.org/10.1016/B978-0-12-801122-5.00009-X

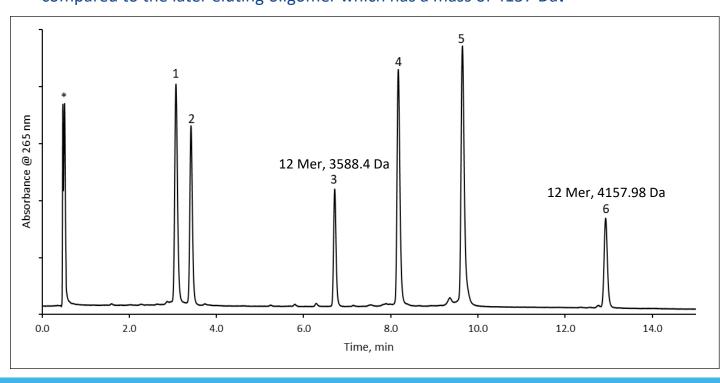


Oligomer Composition Matters!

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- Two oligomers in this mix are the same base length, 12 mer, but are well resolved The difference in retention can be attributed to a difference in sequence which also affects mass.
- The earlier eluting 12 mer oligonucleotide has a mass of 3588 Da compared to the later eluting oligomer which has a mass of 4157 Da.



$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	PEAK IDENTITIES: 1. 20 mer 2. 15 mer 3. 12 mer 4. 25 mer 5. 33 mer 6. 12 mer
Flow Rate: 0.4 mL/min Back Pressure: 142 bar Temperature: 50 °C Injection: 4 µL of Oligonucleotide Performance Standard Mix, 12-33 NT P/N: PHR8667-1EA Sample Solvent: 10mM Tris HCl/ 1mM EDTA Wavelength: PDA, 265 nm Flow Cell: 1 µL Data Rate: 40 Hz Response Time: 0.05 sec. LC System: Shimadzu Nexera X2	

		Chemical	Molecular	Content
Sequence	Mer	Formula	Weight	nmol/vial
	12	$C_{120}H_{157}N_{24}O_{82}P_{11}$	3588.40	0.8
TTT TTT TTT TTT 3'mod {BtnTg} 1	12	$C_{142}H_{197}N_{27}O_{92}P_{12}S_1$	4157.98	1.0
AGC TGT ACT TTT TTT TTT TTT TTT T	25	$C_{248}H_{320}N_{64}O_{165}P_{24}$	7580.90	1.0
AGC TGT ACT TTT TTT TTT TTT TTT TTT TTT TTT	33	$C_{328}H_{424}N_{80}O_{221}P_{32}$	10014.40	1.0
TGT GAC CAC GTA GAC TGA CT	20	$C_{195}H_{246}N_{75}O_{118}P_{19}$	6117.04	1.0
TCT CTC TCT CTC TCT	15	$C_{143}H_{189}N_{37}O_{96}P_{14}$	4395.90	1.0

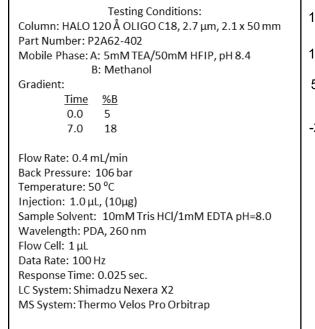
¹ TTT TTT TTT TTT 3'mod {BtnTg} structure below.



LCMS Oligonucleotide N+1/2 Separations

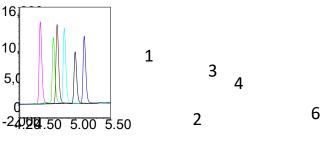
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- Six different oligonucleotides are individually separated on the HALO[®] OLIGO C18 column
- Using overlays of each injection it can be seen that the column has very little trouble separating each oligonucleotide under MS friendly conditions
- Oligonucleotide base length can help predict retention behavior
- Base type also plays a major role in oligomer separations and by using the HALO[®] OLIGO C18 column even minor changes in base type or length can be separated



MS CONDITIONS:

Detection: (-) HESI Spray Voltage: 2.5 kV Sheath gas: 35 Aux gas: 10 Capillary temp: 350 °C Source Heater temp: 300 °C S lens: 60 microscan: 1 max ion time: 200 ms



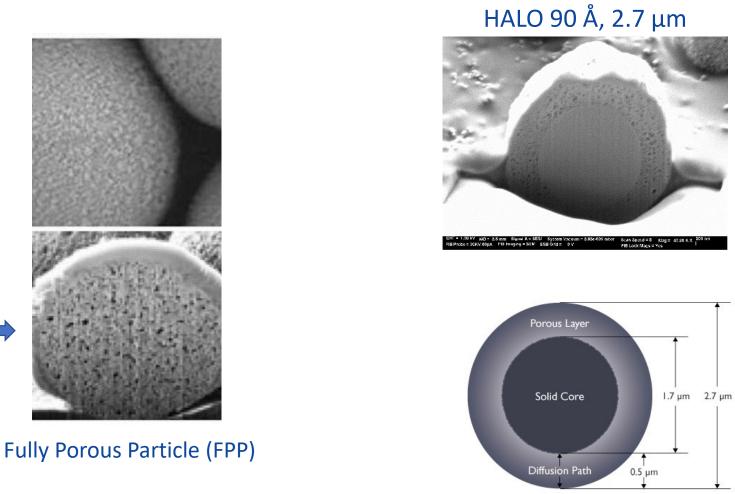
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PE	AK IDENTITIES:	
1.	19 mer	
2.	20 mer	
3.	20 mer	
4.	21 mer	
5.	15 mer	
6.	16 mer	



SPP Technology

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Brief terminology lesson:

FPP – Fully Porous Particle

TPP- Totally Porous Particle

SPP – Superficially Porous Particle

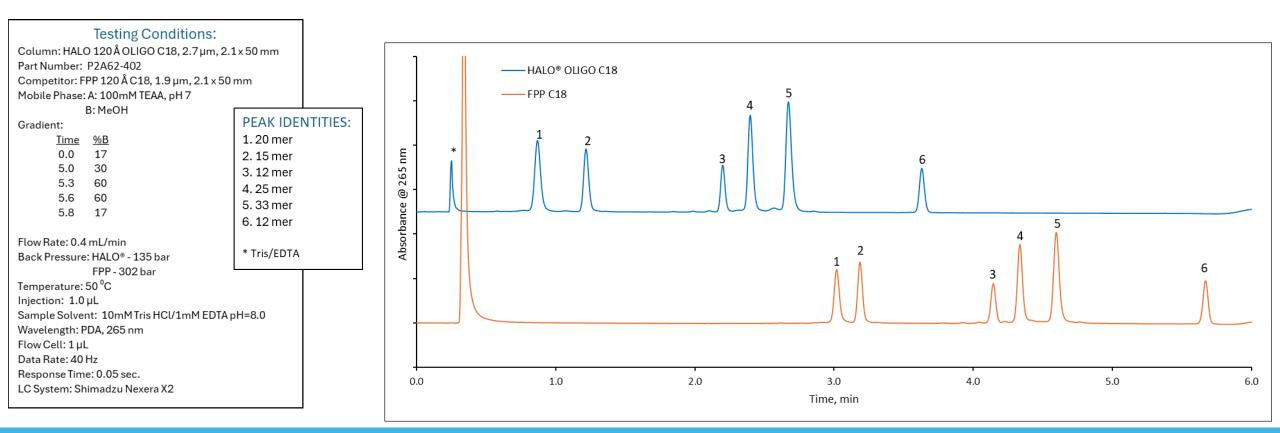
Terms associated with SPP – Fused-Core[®], solid-core, core shell

Superficially Porous Particle (SPP)



Effects of SPP Technology

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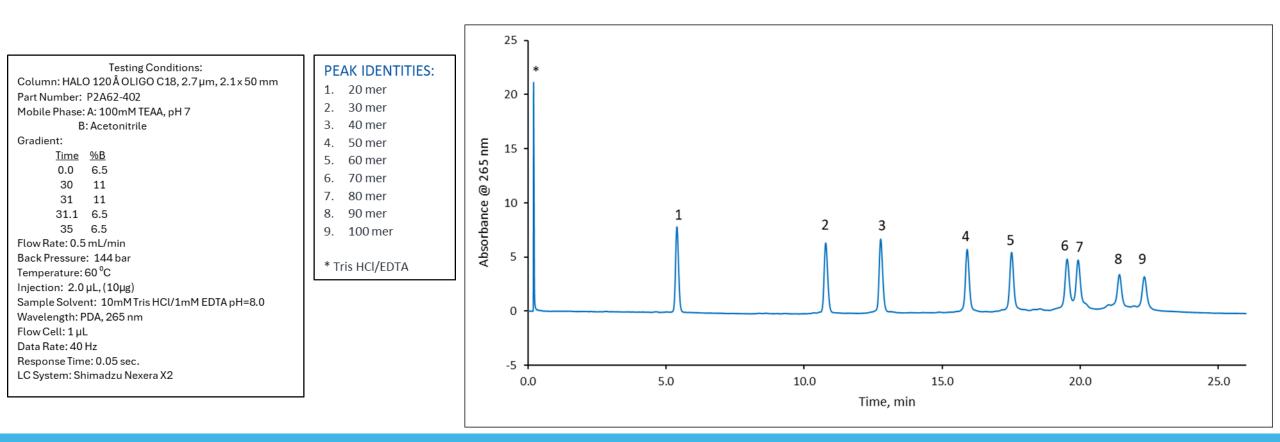




Pore Size Effects on ssDNA

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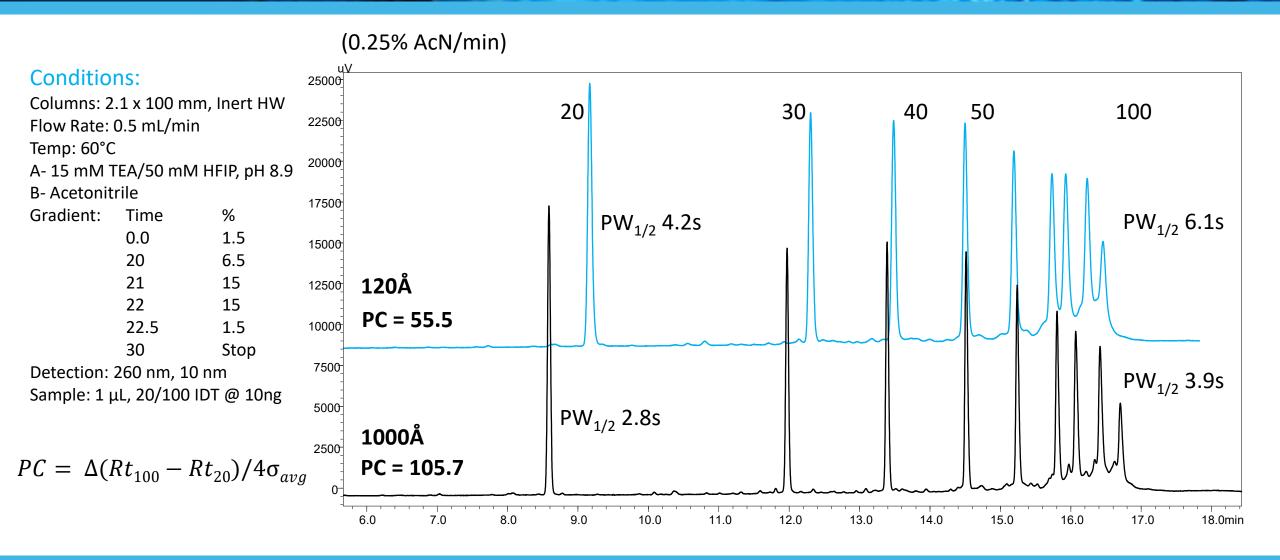
- Baseline separation on all peaks but 6 and 7
- Reaching the limits of a 120Å pore size
- Does a larger pore size increase separation between peaks 6 and 7?





Greater Peak Capacity for Larger Pore SPP: TEA/HFIP Acetonitrile

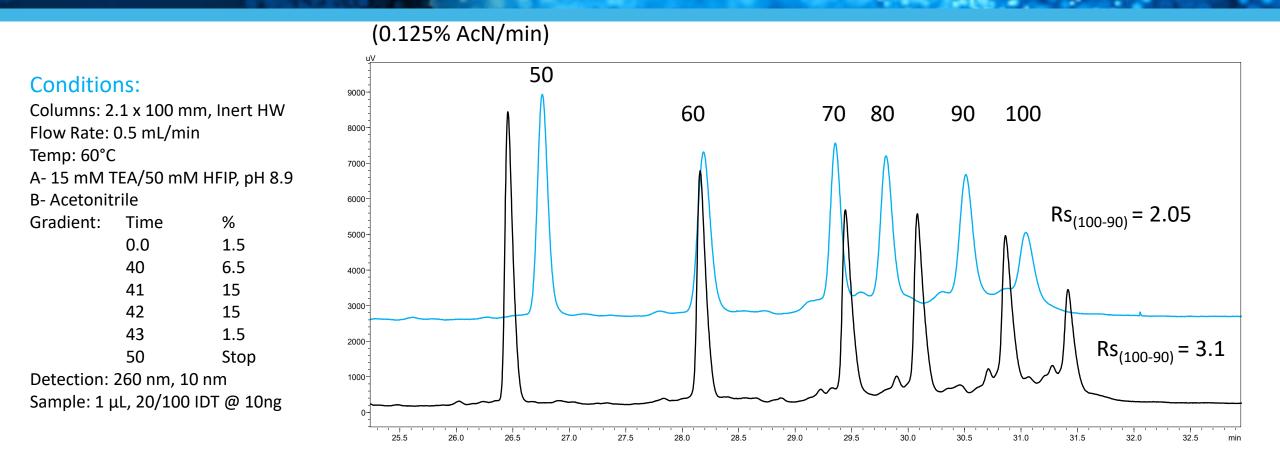
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Resolution of Longer Oligonucleotides

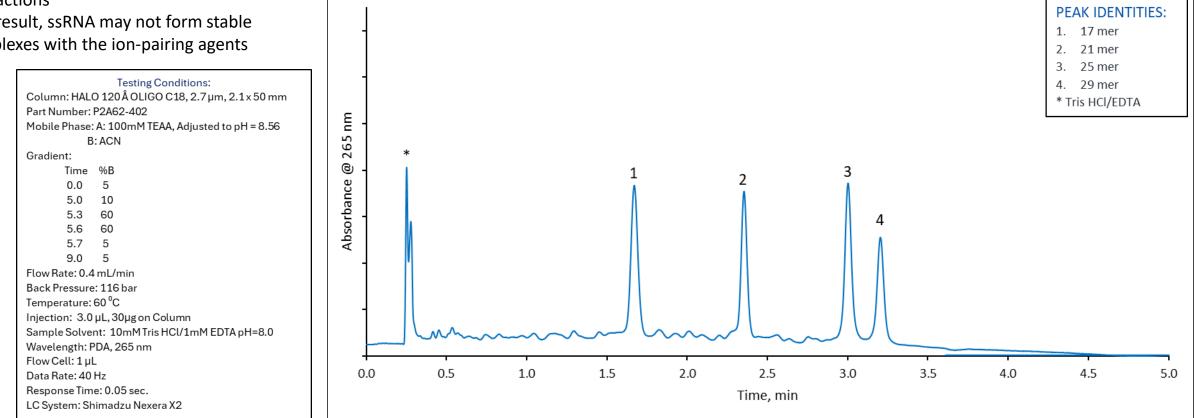
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• Adjusting gradient for closer matching of Rt across this sample yields little effect on Rs or PC

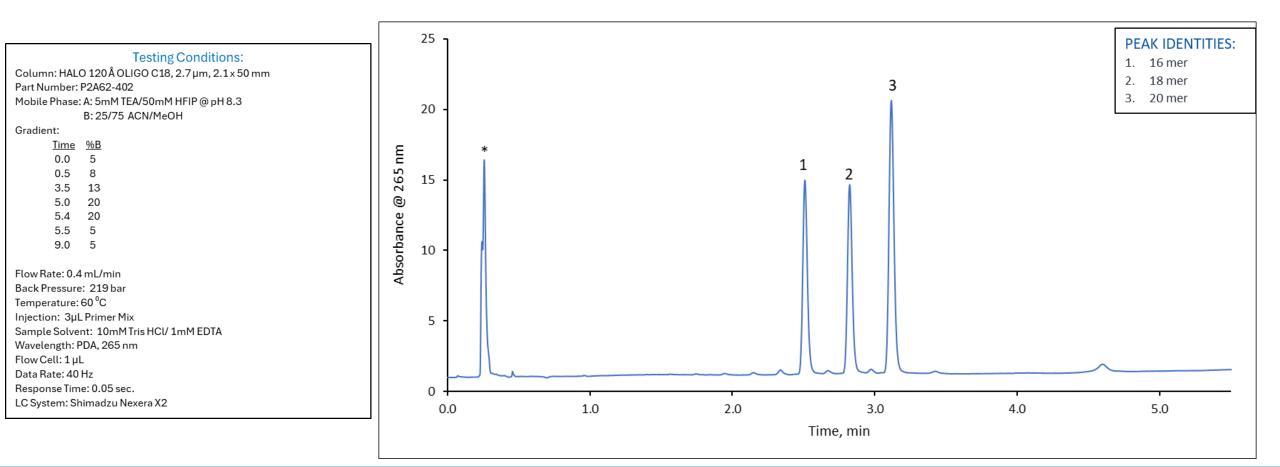
ssRNA Under Ion Pairing

- ssRNA has less retention compared to ssDNA under the same conditions
- ssRNA has a relatively flexible structure that can reduce its hydrophobic interactions
- As a result, ssRNA may not form stable complexes with the ion-pairing agents



Primer Mix on OLIGO C18

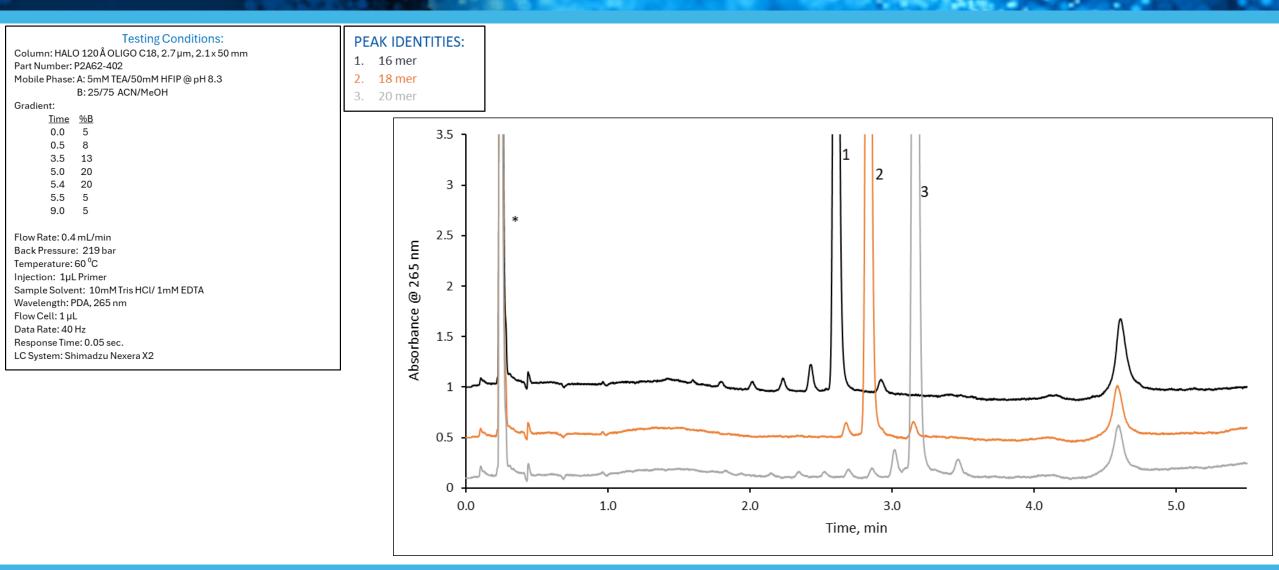
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Primer Separated on OLIGO C18

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Questions?

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Photo by Jamie Street on Unsplash

